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Asymmetric Synthesis. XXXIX.¹ Synthesis of 3-substituted Piperidin-2-ones from Chiral Non-Racemic Lactams.

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Abstract: A series of 3-substituted piperidines in enantiomerically pure form has been synthesized from lactam 1 via the bromo derivative 21. t-Butyl acetate and 5-methylpyridine derivatives 2g and 2h were obtained optically pure by direct alkylation of 1 with corresponding halides. Azido, amino or benzyloxy products were obtained by diastereoselective substitution of 2i.

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INTRODUCTION

In our previous papers ² we reported the diastereoselective alkylation of lactams derived from R-(-)-phenylglycinol (scheme 1). This methodology allowed the preparation of differently substituted piperidines and piperazines. However the reported examples only dealt with non functionalized electrophiles (MeI, PhCH₂Br, CH₂ = CHCH₂ Br). In order to extend the scope of our strategy we decided to investigate the reactivity of lactam 1^{2a} towards heterosubstituted reagents. Our goal was to prepare derivatives 2 bearing amino, alcohol or carboxylic acid functions at C-3 (R = NHR', N₃, OH, CO₂R',COHNR',...) to provide a general access to biologically interesting products.³

Ph₁ OH OH
$$\frac{1}{2}$$
 Base $\frac{1}{2}$ E $\frac{1}{2}$ NBoc

Scheme 1

RESULTS AND DISCUSSION

Lactam 1 was first prepared by our previously described method^{2d}, but recently we observed that it was possible to obtain this compound in 45% yield by condensation of (R)-phenylglycinol with methyl 5-bromovalerate in EtOH in the presence of DBU (see experimental). A first series of experiments was conducted

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with electrophiles generally used for the introduction of a potential amino function (table 1). When the enolate of 1 was reacted with di-tert-butyl azodicarboxylate (DBAD)⁴ (entry 1), no substitution product was observed. Reaction with trisyl azide (TrsN3)⁵ as the electrophile led to azide derivative 2b though in poor yield (10%) as a 1/1 mixture of two C-3 epimers (entry 2). Reaction of the enolate of 1 with Davis's oxaziridine⁶ led to a complex mixture of products in which it has been impossible to detect the expected hydroxyl derivative.

We then turned our attention to the introduction of a potential carboxylic acid function. Reaction of the enolate of 1 with CO₂ (g) furnished acid 2c in quantitative yield but as a 1/1 mixture of two inseparable diastereomeric compounds (entry 3). Reaction with methyl chloroformate led to a complex mixture of C-3 substituted and non substituted carbonates; replacement of s-BuLi by NaHMDS as the base did neither improve the yield of substituted product nor favour the formation of one isomer (entries 4 and 5). The substituted products were also isolated after reaction with tosylcyanide and phenyl isocyanate in 69 % and 52 % yield respectively, however without any diastereoselectivity (entries 6 and 7). In order to determine the origin of the lack of diastereoselection, methyl derivative 3^{2a} was alkylated with phenyl isocyanate (scheme 2). A 1/1 mixture of epimers was also isolated indicating that the absence of diastereoselection was not due to a C-3 epimerization of compounds 2 after alkylation.

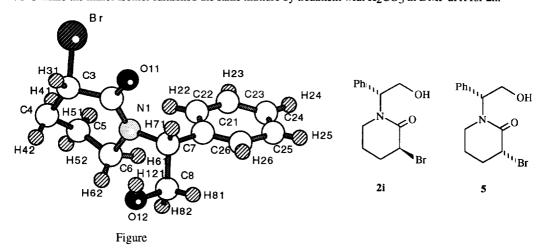
Entry	Base	El	R	Product	Yield (%)	de (%)
1	s-BuLi	DBAD	(NCbzNHCbz)	2a	0	-
2	s-BuLi	TrsN3	N ₃	2 b	10	0
3	s-BuLi	CO ₂	CO ₂ H	2 c	100	0
4	s-BuLi	ClCO ₂ Me	CO ₂ Me	2d	50	0
				(+ carbonate)		
5	NaHMDS	ClCO ₂ Me	-	only carbonate	-	-
6	s-BuLi	TsCN	CN	2 <u>e</u>	69	0
7	s-BuLi	PhNCO	C(O)NHPh	2 f	52	0
8	s-BuLi	BrCH2CO2t-Bu	CH2CO2t-Bu	2 g	44	≥ 95
9	s-BuLi	5-chloromethyl-	(2-chloro-5	2h	69	≥ 95
		2-chloropyridine	pyridinyl)-methyl			
10	s-BuLi (-76°C)	Br2	Br	2i	83	60
11	s-BuLi (-100°C)	Br2	Br	2i	69	90

Table 1: alkylation of lactam 1 with functionalized electrophiles.

Scheme 2

When the functionalization of the electrophile was not located at the reacting center, an excellent diastereoselectivity was observed. Thus, alkylation of lactam 1 with tert-butyl bromoacetate in standard conditions (entry 8) led to compound 2g in moderate yield (44 %) but as a single diastereomer detectable in NMR. By analogy with previous results $2 ext{ a } (R)$ configuration was attributed to the newly created center. When 5-chloromethyl-2-chloropyridine $7 ext{ was used as the electrophile, methylchloropyridine derivative } 2h ext{ was also obtained in good yield (69%) and without any trace of C-3 epimeric compound (entry 9).}$

As the direct functionalization of C-3 position proved to be problematic, we decided to investigate a two-step process involving bromination then nucleophilic substitution. For this purpose the enolate of 1 was reacted with bromine (entries 10 and 11). When the reaction was performed at -100°C, compounds 2i and 5 were isolated in 69 % yield in a 95/5 ratio; at - 70°C a 80/20 ratio was observed. The use of NBS did not improve the diastereoselectivity. The two isomers can be easily separated. An X-ray analysis of the major isomer 2i furnished the C-3 configuration 8 (Figure). The configuration of this asymmetric center was the same as the 3-alkylated piperidin-2-ones previously synthesized by the same method. It is noteworthy that major isomer 5 slowly epimerized to a 80/20 thermodynamic mixture analogous to the reacting mixture obtained at -78°C while the minor isomer furnished the same mixture by treatment with K2CO3 in DMF at rt for 2h.



Having in hand a method providing optically pure bromo derivative 2i on a multigram scale we decided to study the reactivity of this compound towards different nucleophiles (Table 2).

Bromo derivative 2i reacted with NaN3 in CH3CN in the presence of crown ether (18-C-6) to give 6a in 63 % yield and 81 % de. The major isomer can be obtained optically pure after chromatography on silicagel. Although it has been impossible to obtain 6a as a crystalline product suitable for X-ray analysis, we assumed that the reaction occurred with complete inversion of configuration. Cyano derivative 6b was obtained in 71 % yield as an inseparable mixture of two isomers. It is likely that in this case a rapid epimerisation of C-3 due to H-3 acidity could explain this result.

Entry	Nucleophilic	Product	R	Yield (%)	de (%)
1	NaN3, 18-C-6	6a	N3	63	81
2	Bu4N+CN-	6b	CN	71	0
3	PhCH ₂ NH ₂	6c	NHCH2Ph	95	80
4	PhCH ₂ OH	6d	OCH ₂ Ph	46	≥ 95

Table 2: Substitution of bromo derivative 2i with nucleophiles.

Benzylamine reacted with 2i in CH₃CN at rt (entry 3) furnishing derivative 6c in 95 % yield and with good diastereoselectivity (de: 80%). Hydrogenolysis of 6c (H₂, Pd/C) quantitatively furnished 3-amino-piperidin-2-one 7 (scheme 3). Unfortunately, all our attempts to obtain 6c and 7 as single isomers failed. These products were always contaminated by 10-15% of C-3 isomer. However pure amino compound 7 could be obtained independently by reduction of azide 6a. 11

$$\begin{array}{c|ccccc} Ph_{1_{1_{1}}} & OH & Ph_{1_{1_{1}}} & OH \\ \hline N & O & \hline H_{2} & \hline N & O \\ \hline N & Ph & MeOH & \hline N & O \\ \hline 6c & H & 7 & \hline \end{array}$$

Scheme 3

In order to provide C-3 O-substituted derivatives, bromo derivative 2i was reacted with benzyl alcohol. The reaction was complete after 2 h at rt in DMF, after formation of the alcoholate (NaH). A single isomer 6d was obtained in 46 % yield.

The model proposed to explain the diastereoselectivity observed with lactam 1 involves an intramolecular chelation (scheme 4)^{2a} in which the reacting intermediates (A or B) are rigidified thereby allowing a facial differentiation during the attack.

During this work, we observed no diastereoselection with some electrophiles which possess an electron rich function (Lewis base) in the vicinity of the reacting center of the electrophile. Presumably this leads to an intermolecular chelation process with the lithium atom of the alcoholate which would modify the geometry of the intermediate, and therefore the diastereoselectivity. If this function is separated from the reacting center by at least one non substituted carbon (e.g. BrCH₂CO₂tBu) diastereoselectivity was then recovered.

EXPERIMENTAL SECTION

¹H NMR and ¹³C NMR spectra were recorded at 300 MHz and 75 MHz respectively on a Bruker AC 300 spectrometer. Optical rotations were measured with a Perkin-Elmer 241 automatic polarimeter. Product purification was performed by flash chromatography on Silica Gel (Merck 60). All reactions involving air sensitive materials were carried out under a N₂ atmosphere. Compound **2d** which has been obtained as an inseparable mixture of diastereoisomers and carbonates is not described.

(αR) -N-(2-Hydroxy-1-phenylethyl)piperidine-2-one 1:

A solution of (R)-(-)-phenylglycinol (1.37 g, 10⁻² mol), methyl 5-bromovalerate (1.95g, 10⁻² mol) and DBU (3.04g, 2 10⁻² mol) in EtOH (20 mL) was stirred at reflux for 48 h. After evaporation of the solvent under reduced pressure, the residue was diluted with CH₂Cl₂ (50 mL) then washed with a saturated aqueous solution of NH₄Cl. The organic layer was dried over MgSO₄ then evaporated furnishing an oil which was purified by flash-chromatography (CH₂Cl₂/MeOH: 95/5). Lactam 1 was obtained as an oil (985 mg, 45%) which slowly crystallized. For physical and spectroscopic data see ref. 2d.

General procedure for preparation of compounds 2c-h:

To a solution of lactam 1 in THF (50 mL/1g of 1) under nitrogen, was added s-BuLi (2.5 eq) at -78°C. The mixture was stirred for 20 min, and electrophile (2.5 eq) in minimum of THF was added dropwise. After stirring at -78°C for 3 h, the mixture was treated with saturated aqueous solution of NH4Cl, extracted twice

with AcOEt, dried over MgSO4 and concentrated. The crude product was then purified by flash chomatography.

$(\alpha R, 3R)$ and $(\alpha R, 3S)$ -[N-(2-hydroxy-1-phenylethyl)piperidin-2-one-3-yl]carboxylic acid 2c:

Reaction of enolate of 1 with CO₂ (gas) furnished an inseparable mixture of epimeric acids in a 1/1 ratio. 1 H NMR (δ ppm): 1.45-2.10 (m, 8H), 2.23 (m, 2xH-3), 2.92-3.50 (m, 4xH-6), 4.10-4.20 (m, 4H, 4xH-8), 5.82 (t, J = 6.3 Hz, H-7), 5.93 (t, J = 6.3 Hz, H-7), 7.10-7.40 (m, 10H). 13 C NMR (δ ppm): 21.2, 21.5, 23.2, 23.4 (C-4 and C-5), 43.1, 43.7 (C-6), 46.1, 46.4 (C-3), 58.5, 59.0 (C-7), 60.5 (2xC-8), 127.6, 127.8, 128.3, 129.0, 135.6, 135.9, 170.5 and 171.6 (C-2), 182.3 (\underline{C} O₂H). IR (cm⁻¹): 3390, 1720; 1620. MS (CI): 264 (MH⁺).

$(\alpha R, 3R)$ and $(\alpha R, 3S)$ -3-cyano-N-(2-hydroxy-1-phenylethyl)piperidin-2-one 2e :

Cyano derivative 2e was obtained as a 1/1 mixture of two epimeric compounds. 13 C NMR (δ ppm): 19.7, 20.9, 23.9, 25.7 (C-4 and C-5), 35.3, 35.7 (C-3), 42.6, 42.8 (C-6), 58.5, 58.9 (C-8), 60.6, 60.7 (C-7), 118.2 (CN), 127.8, 128.2, 128.7, 128.9, 136.0, 171.6 (C-2).

$(\alpha R, 3R)$ and $(\alpha R, 3S)$ -N-(2-hydroxy-1-phenylethyl)-3-benzamidopiperidin-2-one 2f :

Benzamide 2f was obtained as a 1/1 mixture of two epimeric compounds in 52% yield. 1 H NMR (δ ppm) : 1.6-2.3 (m, 8H), 2.90-3.25 (m, 4H, 4xH-6), 3.50 (m, 2xH-3), 4.10-4.20 (m, 4xH-8), 5.80-5.90 (m, 2xH-7), 7.05-7.55 (m), 9.6-9.7 (br.s, NH). 13 C NMR (δ ppm) : 21.0, 21.3, 23.1, 23.2 (C-4, C-5), 43.5, 43.7 (C-6), 48.8, 48.9 (C-3), 58.4, 58.9 (C-7), 60.8, 61.2 (C-8), 119.9, 120.0, 124.1, 124.2, 127.5, 127.8, 128.0, 128.2, 128.7, 128.9, 136.3, 138.1, 166.8 and 167.0 (CONHPh), 170.1 (C-2). IR (cm $^{-1}$) : 3320, 1682, 1614. MS (EI) : 338, 320, 308, 77.

$(\alpha R, 3R)$ -3-tert-Butyloxycarbonylmethyl-N-(2-hydroxy-1-phenylethyl)piperidin-2-one 2g :

2g was obtained in 44% yield as a colourless oil. [α]D = -72 (c : 1.07; EtOH). ¹H NMR (δ ppm): 1.45 (s, 9H, 3 Me), 1.61-1.70 (m, 1H, H-4), 1.70-1.86 (m, 2H, 2 H-5), 1.88-1.97 (m, 1H, H-4), 2.68-2.80 (m, 3H, 2 H-9, H-3), 3.12 (m, 1H, H-6), 3.22 (m, 1H, H-6), 4.1 (dd, J = 11.6, 4.8 Hz, 1H, H-8), 4.28 (dd, J = 11.6, 9.6 Hz, 1H, H-8), 5.31 (dd, J = 9.6, 4.8, 1H, H-7), 7.23-7.38 (m, 5H ar.). ¹³C NMR (δ ppm): 22.6 (C-4), 26.3 (C-5), 28.2 (3 Me), 37.6 (C-9), 39.2 (C-3), 46.4 (C-6), 62.5 (C-7, C-8), 80.9 (C-11), 127.8, 127.9, 128.7 (C ar.), 137.3 (C ar. quat.), 172.0 (C-10), 173.4 (C-2). IR (cm⁻¹): 3409; 1720; 1621. MS (CI): 334 (MH⁺). Anal. Calcd for C₁9H₂7NO₄: C, 68.44; H, 8.16; N, 4.20. Found: C, 68.78, H, 7.96, N, 4.31.

(\alpha R, 3R)-3-(2-Chloropyridin-5-yl-methyl)-N-(2-hydroxy-1-phenylethyl)piperidin-2-one 2h :

2h was obtained in 69% yield as a colourless oil. $[\alpha]_D = -22$ (c: 1.09, EtOH). ¹H NMR (δ ppm): 1.43 (m, 1H, H-4), 1.6-1.8 (m, 3H), 2.68-2.82 (m, 2H, H-3 and H-6), 2.91 (dd, J=13.8, 8.4, 1H, H-6), 3.21 (m, 2H, 2 H-9), 3.80 (br. s, 1H, OH), 4.03-4.18 (m, 2H, 2 H-8), 5.86 (dd, J = 8.8, 5.8 Hz, 1H, H-7), 7.18 (m, 2H ar), 7.21 (d, J = 8.1 Hz, H-3'), 7.27-7.35 (m, 3H ar), 7.53 (dd, J = 8.1, 2.5 Hz, 1H, H-4'), 8.21 (d, J = 2.5 Hz, 1H, H-6'). ¹³C NMR (δ ppm): 21.5, 25.2 (C-4 and C-5), 34.2 (C-9), 43.0 (C-3), 43.4 (C-6), 58.4 (C-7), 61.2 (C-8), 124.0 (C-2'), 128.7 (C ar), 134.1 (C quat. ar), 136.8 (C-5'), 139.9 (C-4'), 149.4 (C-2'), 150.4 (C-6'), 172.7 (C-2). IR (cm⁻¹): 3394, 3018, 1619, 1215. MS (CI): 346, 344. Anal. Calcd for C₁₉H₂₁N₂O₂Cl + 1/2 A₂OEt: C, 64.86, H: 6.48, N: 7.20. Found: C, 65.41, H, 6.66, N, 7.42.

$(\alpha R, 3S)$ -3-Bromo-N-(2-hydroxy-1-phenylethyl)piperidin-2-one 2i and $(\alpha R, 3R)$ -3-Bromo-N-(2-hydroxy-1-phenylethyl)piperidin-2-one 5:

To a solution of lactam 1 (1.2 g, 5.4 mmol) in THF (100 ml) under nitrogen was added s.BuLi (2.5 eq.) at -78°C. After being stirred for 20 min, the mixture temperature was lowered to -100°C and bromine (288 μl, 5.4 mmol) was added dropwise. After being stirred at -100°C for 5 min, the mixture was treated with saturated NH4Cl, extracted twice with AcOEt, dried over MgSO4 and concentrated. The crude product was then roughly purified by flash chomatography (AcOEt) to give the bromo derivative as a 95/5 diastereomeric mixture. Optically pure 2i (1.02g, 65%) and 5 (64 mg, 4%) were then obtained after a second flash

chromatography (AcOEt). The same reaction performed at -76°C led to 2i and 5 in 66% and 17% yield respectively.

2i [α]_D= -112 (c : 0.8 ; MeOH). mp = 129°C (ethyl acetate). ¹H NMR (δ ppm): 1.71 (m, 1H, H-5) ; 2.11-2.38 (m, 3 H, 2 H-4, H-5) ; 2.61 (s, 1H, OH) ; 3.12 (m, 1H, H-6 eq.) ; 3.32 (td, J = 6.4, 4.7 Hz, 1H, H-6 ax.) ; 4.12 (m, 1H, H-8) ; 4.22 (dd, J = 5.1, 5.0 Hz, 1H, H-8) ; 4.68 (td, J = 3.3, 1.3 Hz, 1H, H-3) ; 5.85 (dd, J = 5.1, 4.3 Hz, 1H, H-7) ; 7.20 - 7.40 (m, 5H, ar.). ¹³C NMR (δ ppm): 18.6 ; 30.9 (C 4, C 5) ; 42.2 (C 6) ; 46.0 (C 3) ; 57.3 (C 7) ; 60.2 (C 8) ; 127.2 ; 127.6 ; 128.6 (C ar.) ; 136.5 (C ar. quat.) ; 168.0 (C 2). IR (cm⁻¹) : 3382 ; 1631. MS (EI) : 300 ; 298 (MH⁺⁺) ; 268 ; 266 ; 200 ; 188 ; 186 ; 180 ; 178 ; 159 ; 130 ; 121 ; 104 : 86 ; 84 ; 70. HRMS calcd. for 298.0442 (M+ H⁺) : found : 298.0450.

 $5 \text{ [}\alpha\text{]}_D = -104 \text{ (c}: 1.1, \text{MeOH)}. \text{ Amorphous.} \ ^{1}\text{H} \text{ NMR (}\delta\text{ ppm)} \ 1.60 \text{ (m, 1H)}, 2.20 \text{ (m, 3H)}, 3.05 \text{ (m, 1H, H-6 eq)}, 3.30 \text{ (m, 2H, H-6ax and OH)}, 4.12 \text{ (m, 2H, 2 H-8)}, 4.63 \text{ (td, J} = 4.4, 1.2 \text{ Hz, H-3)}, 5.53 \text{ (dd, J} = 8.1, 5.8 \text{ Hz, H-8)}, 7.20-7.40 \text{ (m, 5H ar)}. \ ^{13}\text{C} \text{ NMR (}\delta\text{ ppm)}: 19.5, 31.1 \text{ (C-4 and C-5)}, 44.3 \text{ (C-6)}, 45.9 \text{ (C-3)}, 60.3 \text{ (C-8)}, 61.8 \text{ (C-7)}, 128.0, 128.9 \text{ (CH ar)}, 136.4 \text{ (Cq ar)}, 168.1 \text{ (C-2)}. \text{ IR (cm$^{-1}$)}: 3395, 1635. \text{MS (EI)}: 300, 298 \text{ (MH$^{+}$)}.$

$(\alpha R, 3R)$ -3-Azido-N-(2-hydroxy-1-phenylethyl)piperidin-2-one 6a :

To a solution of bromo lactam 2i (70 mg; 0.24 mmol) in anhydrous acetonitrile (5 ml) was added sodium azide (17 mg, 0.26 mmol) and a catalytic amount of crown 18-C-6 (4 mg). The mixture was stirred at rt for 2 h, filtered and dried over MgSO4. The solvent was evaporated and the crude product was roughly purified by flash chromatography (AcOEt) to give 6a (38 mg, 63% yield) as a 90/10 diastereomeric mixture. Optically pure 6a was then obtained after a second flash chromatography (AcOEt) as a colourless oil (27 mg, 44% yield). [α]D = + 30 (c: 0.75; CHCl3). ¹H NMR (δ ppm): 1.61-2.09 (m, 4H, 2 H-5, 2 H-4), 2.95 (m, 1H, H-6), 3.19 (m, 1H, H-6), 4.12 (m, 3H, 2 H-8, H3), 5.73 (dd, J = 8.5, 6.0 Hz, 1 H, H-7), 7.12-7.36 (m, 5 H ar.). ¹³C NMR (δ ppm): 20.4, 27.4 (C-4, C-5), 43.4 (C-6), 59.1 (C-7), 59.9 (C-3), 61.4 (C-8), 127.9, 128.2, 128.9 (C ar.), 136.4 (Cq ar.), 169.3 (C-2). IR (cm⁻¹): 3426; 2103 (N₃); 1635; 1443. MS (EI): 261 (MH+*), 242, 229, 200, 121, 103, 91, 77. HRMS calcd. for 261.1352 (M+ H+*); found: 261.1364.

$(\alpha R, 3S)$ -3-Benzylamino-N-(2-hydroxy-1-phenylethyl)piperidin-2-one 6c:

To a solution of bromo lactam 2i (104 mg; 0.35 mmol) in anhydrous acetonitrile (3 ml) was added benzylamine (76 μ l, 0.27 mmol). The mixture was stirred at rt for 12 h, treated with aqueous potassium carbonate extracted twice with CH₂Cl₂ and dried over MgSO₄. Solvant was evaporated and the crude product was purified by flash chromatography (CH₂Cl₂ / MeOH 98:2) to give 6c (70 mg, 62% yield) as a 90/10 diastereomeric mixture. Major diastereoisomer: 1 H NMR (δ ppm) (CD₃OD): 1.62-2.32 (m, 4H, 2 H-4, 2 H-5), 3.13 (m, 1H, H-3), 3.82 (d, J = 13.5 Hz, 1H, H-9), 3.95 (d, J = 13.5 Hz, 1H, H-9), 4.11-4.23 (m, 2H, 2 H-8), 5.91 (m, 1H, H-7), 7.31-7.52 (m, 10H ar.). 13 C NMR (δ ppm) (CD₃OD): 21.9, 27.1 (C-4, C-5), 43.7 (C-6), 51.7 (C-9), 58.5 (C-3), 59.2 (C-7), 61.1 (C-8), 128.4, 128.6, 128.8, 128.9, 129.6 (CH ar.), 138.6, 140.4 (Cq ar.), 173.1 (C-2). MS (CI): 325 (MH⁺).

$(\alpha R, 3S)$ -3-Benzyloxy-N-(2-hydroxy-1-phenylethyl)piperidin-2-one 6d:

To a suspension of NaH (29 mg; 1.2 mmol) in DMF (3 mL) was added benzyl alcohol (130 mg; 1.2 mmol) at rt. After stirring for 20 min, a solution of bromo lactam 2i (180 mg; 0.6 mmol) in DMF (3 mL) was added slowly. Stirring was maintained at rt for 1 h, then a saturated solution of NH4Cl (10 mL) was added. The mixture was extracted with Et₂O. Organic layer was dried over MgSO₄ then evaporated. Crude product was purified by flash-chromatography (AcOEt / cyclohexane: 90/10) furnishing 6d (90 mg; 46% yield) as a colourless oil. $[\alpha]_D = -28$ (c: 1.0, MeOH). ¹H NMR (δ ppm): 1.55-2.85 (m, 4H, 2 H-5 and 2 H-4), 3.21 (m, 1H, H-6eq), 3.45 (m, 1H, H-6ax), 3.82 (t, J = 5.9 Hz, 1H, H-3), 4.05 (m, 2H, 2 H-8), 4.71 and 4.85 (2 d, J = 11.9 Hz, 2H, OCH₂Ph), 5.68 (dd, J = 8.0, 5.9 Hz, 1H, H-7), 7.15-7.40 (m, 5H ar). ¹³C NMR (δ ppm) 19.7, 27.6 (C-4 and C-5), 43.0 (C-6), 58.7 (C-3), 61.5 (OCH₂Ph), 72.8 (C-8), 75.1 (C-7), 126.9, 128.6 (CH ar), 136.9, 138.3 (C quat. ar), 171.3 (C-2). IR (cm⁻¹): 3402, 1635, 1453. MS (EI): 325 (1), 294 (15), 266 (15), 219 (100), 188 (100), 91 (100). Anal. Calcd for C₂0H₂3NO₃: C, 73.82, H, 7.12, N, 4.30. Found: C, 73.59, H, 7.31, N, 4.34.

$(\alpha R, 3S)$ -3-Amino-N-(2-hydroxy-1-phenylethyl)piperidin-2-one 7:

A solution of benzylamine **6c** (90/10 diastereomeric mixture) (572 mg, 1.76 mmol) in MeOH (5 mL) was stirred for 48h under H₂ in the presence of Pd(OH)₂ (10 mg). After the catalyst was removed by filtration over celite and the solvent was evaporated, primary amine 7 (413 mg, 100% yield) was obtained as a 90/10 diastereomeric mixture. Major isomer: 1 H NMR (CD₃OD) (8 ppm): 1.9-2.4 (m, 4H, 2 H-4 and 2 H-5), 3.25 (dtd, J = 12.5, 4.5 1.0 Hz, 1H, H-6eq), 3.51 (ddd, J = 12.5, 10.0, 4.6 Hz, 1H, H-6ax), 4.10 (dd, J = 11.5, 6.2 Hz, 1H, H-3), 4.20 (m, 2H, 2 H-8), 5.85 (dd, J = 7.7, 7.3 Hz, 1H, H-7), 7.3-7.5 (m, 5H ar). 13 C NMR (CD₃OD) (8 ppm): 20.8, 25.3 (C-4 and C-5), 41.8 (C-6), 51.2 (C-3), 57.8 (C-7), 59.8 (C-8), 127.6, 128.0, 128.8 (CH ar), 136.4 (C quat ar), 168.0 (C-2). IR (cm⁻¹): 3376, 1649. MS (CI): 235.

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- 8. Crystal data: C13H16NBr02. $M_w = 298.19$. A suitable crystal was investigated on a Siemens P3 diffractometer ($\lambda(\text{MoK}\alpha) = 0.71069$ Å, graphite monochromator). Orthorhombic, space group P212121, Z=4, a=8.995(4), b=9.8445(5), c=14.267(8), d_c = 1.57g.cm⁻³, F(000) = 310, μ = 0.31 mm⁻¹. 2118 reflections up to 20 = 60° of wich 1360 with I> 3 $\sigma(\text{I})$ were kept in refinement calculations. The structure was solved by direct methods using SHELX86⁹ and refined by least squares with SHELX76¹⁰, minimising the quantity $\Sigma w(\text{Fo-Fc})^2$; all non-hydrogens atoms were located in difference Fourier maps and refined at observed positions with isotropic temperature factors. Final R= $\Sigma(\text{Fo-Fc})/\Sigma\text{Fo} = 0.046$, W_r=($\Sigma w(\text{Fo-Fc})^2/\Sigma w\text{Fo}^2$)1/2=0,046 with w=1/ $\sigma^2(\text{Fo})$.
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